

Copper(II)-Catalyzed Three-Component Arylation/Hydroamination Cascade from Allyl Alcohol: Access to 1-Aryl-2-sulfonylamino-propanes

Camilla Loro,* Marta Papis, Francesca Foschi, Gianluigi Broggin, Giovanni Poli, and Julie Oble*



Cite This: *J. Org. Chem.* 2023, 88, 13995–14003



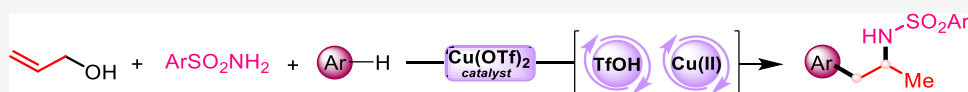
Read Online

ACCESS |

Metrics & More

Article Recommendations

Supporting Information



ABSTRACT: A new straightforward approach to 1-aryl-2-aminopropanes using easily accessible substrates has been developed. Simple allyl alcohol is shown to be an ideal synthetic equivalent of the C3 propane-1,2-dylium bis-cation synthon in three-component cascade reactions with arenes and sulfonamide nucleophiles to regioselectively afford 1-aryl-2-aminopropanes. The reaction is catalyzed by $\text{Cu}(\text{OTf})_2$ and is expected to involve a Friedel–Crafts-type allylation of the arene, followed by hydroamination.

INTRODUCTION

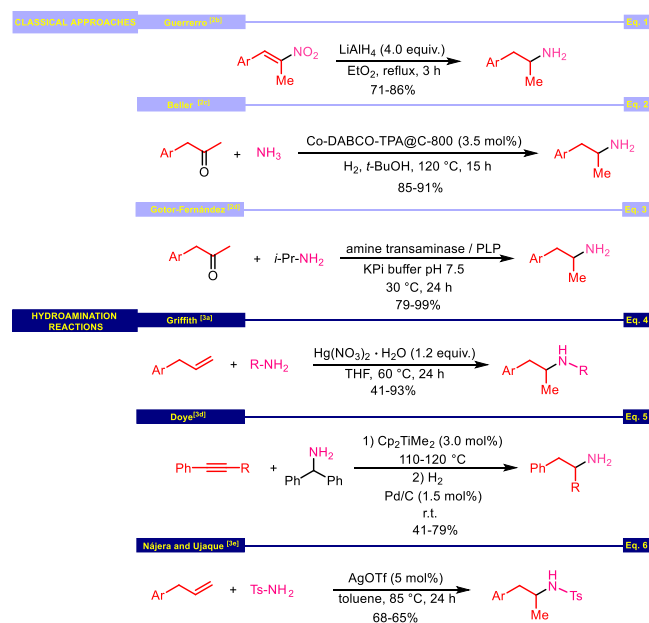
1-Aryl-2-aminopropanes are widely applied in synthetic chemistry and in the pharmacological field. Also known as amphetamines, they are substances that belong to the psychoanaleptic group, known for their stimulating effects on the sympathetic nervous system as well as for their ability to inhibit various enzymes.¹

Several methods for the synthesis of 1-aryl-2-aminopropanes are reported in the literature, many of which are based on classical reactions of organic chemistry (Scheme 1, eqs 1–3).² Complementary methods reach these compounds by hydroamination of allyl, vinyl, or alkynyl arenes promoted by various catalysts or promoters (Scheme 1, eqs 4–6).³

Being interested in transformations involving the concatenated generation of several bonds in a single synthetic operation,⁴ we have recently developed a new copper-promoted reaction, which allows access to 1-aryl-2-aminopropanes starting from *O*-allyl *N*-tosyl carbamates.⁵ This synthetic procedure, although innovative, required the use of a large excess of $\text{Cu}(\text{OTf})_2$. For this reason, we decided to pursue our studies to further upgrade this synthetic transformation.

Allyl alcohol derivatives have been used in various protocols as variously substituted electrophilic C3 synthons through the involvement of either the corresponding π -allyl metal complexes (in the presence of catalytic amounts of low-valent transition metals)⁶ or the corresponding allylic cations (in the presence of a protic acid promoter).⁷ Allylic alcohols have also attracted considerable interest in the field of Friedel–Crafts (FC) reactions, enabling the allylation of aromatic or heteroaromatic systems.⁸ A representative example involves the allylation of electron-rich (hetero)arenes with allylic alcohols, in which the carbocationic species is generated by

Scheme 1. Selected Procedures for the Synthesis of 1-Aryl-2-aminopropanes



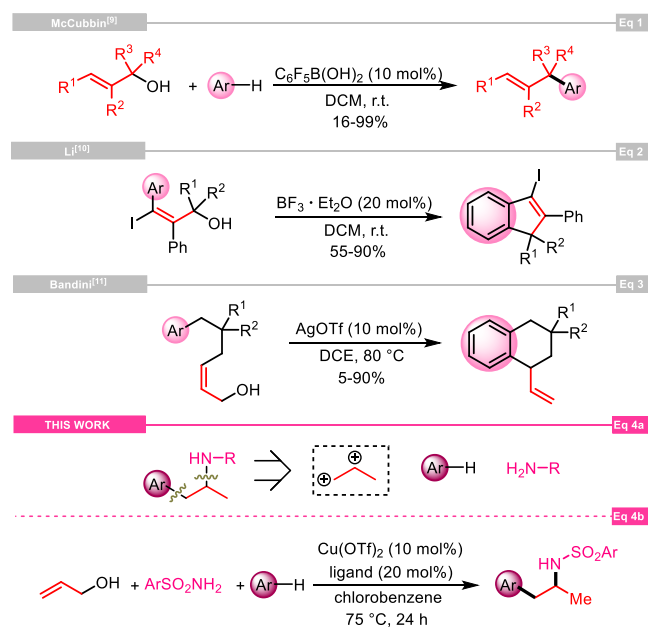
Received: July 10, 2023

Published: September 25, 2023



catalytic amounts of pentafluorophenylboronic acid (Scheme 2, eq 1).⁹ In 2008, an innovative way to synthesize 3-

Scheme 2. Allylic Alcohols as Various Substituted C3 Synthons in Friedel–Crafts Reactions



iodoindenes through an intramolecular FC reaction of 3-iodo-3-arylprop-2-en-1-ols in the presence of catalytic amounts of $F_3B\cdot OEt_2$ was reported (Scheme 2, eq 2).¹⁰ One year later, Bandini and co-workers obtained 1-vinyl-tetrahydronaphthalenes by the cyclization of 6-arylhex-2-en-1-ol motifs with $AgOTf$ (Scheme 2, eq 3).¹¹ As for our contribution, we envisioned the use of the allylic alcohol motif as a synthetic equivalent of the C3 propane-1,2-dylium bis-cation synthon in cascade reactions with aryl derivatives and nitrogen-based nucleophiles to regioselectively reach 1-aryl-2-aminopropanes (Scheme 2, eq 4a). Such a three-component process could be successfully attained using sulfonamides and electron-rich arenes in the presence of catalytic $Cu(OTf)_2$ under very mild conditions (Scheme 2, eq 4b).

RESULTS AND DISCUSSION

Our investigation began with the evaluation of the reaction conditions previously adopted with the *O*-allyl carbamates.⁵ Accordingly, reacting allyl alcohol with 2.0 equiv of tosylamide in the presence of 4.0 equiv. of $Cu(OTf)_2$ in mesitylene as the solvent at 130 °C for 3.0 h gave a mixture of 1,2-arylation/hydroamination **1a** and 1,2-diarylation products **2** in 34 and 23% yields, respectively (Table 1, entry 1). Using chlorobenzene as the solvent and 5 equiv of mesitylene gave selectively the three-component C–C/C–N coupling product **1a** in 63% isolated yield (entry 2). Lowering the amount of $Cu(OTf)_2$ to 1.0 equiv was nearly as effective (entry 3), while using 10 mol % $Cu(OTf)_2$ gave a lower yield for **1a** (entry 4). To improve this result, this copper-catalyzed reaction was studied in the presence of different ligands,¹² and to our delight, the use of the diphosphine ligand xantphos selectively led to **1a** in 78% yield after 24 h at 100 °C (entry 5). By lowering the reaction temperature to 75 °C, the yield of **1a** further increased to 81% (entry 6).¹³ However, further lowering the reaction temperature to 50 °C only returned

Table 1. Optimization of the Reaction Conditions^a

Ar = mesityl

entry	additive (equiv)	ligand (20 mol %)	time (h)	temp. (°C)	yield ^b (%)
1 ^c	$Cu(OTf)_2$ (4.0)		3	130	1a (34) + 2 (23)
2	$Cu(OTf)_2$ (4.0)		4	130	1a (63)
3	$Cu(OTf)_2$ (1.0)		6	130	1a (60)
4	$Cu(OTf)_2$ (0.1)		6	130	1a (49)
5	$Cu(OTf)_2$ (0.1)	xantphos	24	100	1a (78)
6 ^d	$Cu(OTf)_2$ (0.1)	xantphos	24	75	1a (81)
7	$Cu(OTf)_2$ (0.1)	xantphos	24	50	S.M.
8	TfOH (0.05)		4	130	1a (71) + 2 (12)
9	TfOH (0.2)		24	75	1a (trace) + degr. products

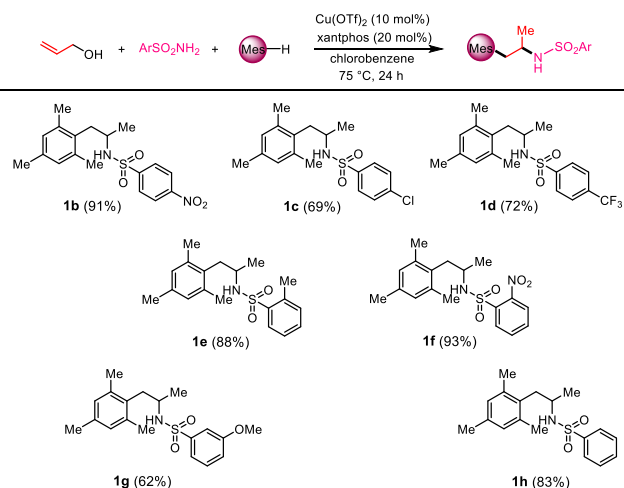
^aReaction conditions: allyl alcohol (1.0 equiv), tosylamide (2.0 equiv), mesitylene (5.0 equiv), and chlorobenzene (0.25 M) at 75 °C in an oil bath for 24 h. ^bIsolated yields. ^cReaction performed in mesitylene as the solvent (0.25 M). ^dReaction performed with 2.0 equiv of mesitylene gave compound **1a** with 35% yield.

the starting substrate back (entry 7), while dropping the amount of mesitylene to 2.0 equiv lowered the yield of **1a** to 35%. Carrying out the coupling in the presence of 5 mol % TfOH, without $Cu(OTf)_2$, gave **1a** and **2** in 71% and 12% isolated yields, respectively, which suggests the *in situ* generation of this acid in the reaction medium (entry 8). Finally, an additional experiment using 20 mol % TfOH at 75 °C afforded only traces of **1a** with a lot of degradation (entry 9).

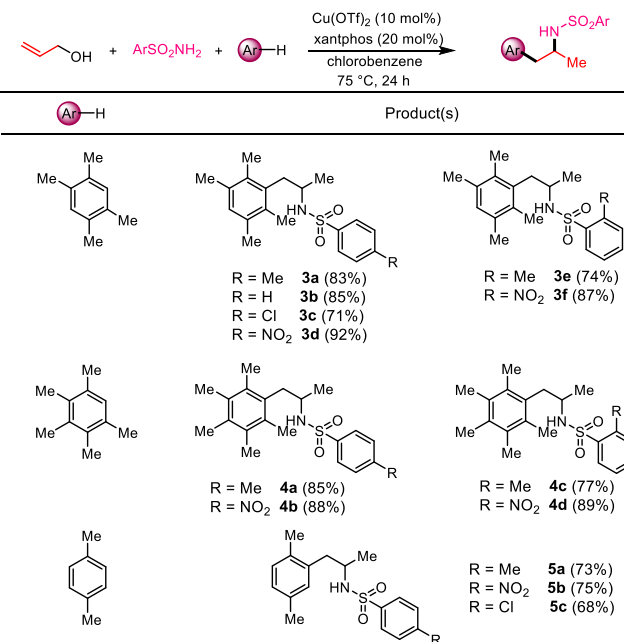
With these optimized conditions in hand, we then proceeded to test this three-component reaction with other sulfonamides (Scheme 3). All of the aryl sulfonamides tested, which incorporated electron-donating and -withdrawing groups on the phenyl ring, gave the expected corresponding products (**1b–h**) in good to excellent yields.

We then explored the substrate scope by using a series of different electron-rich aromatic hydrocarbons and variously substituted aromatic *N*-sulfonamides (Scheme 4). Gratifyingly, durene, 1,2,3,4,5-pentamethylbenzene, and *p*-xylene gave the expected three-component coupling products **3a–f**, **4a–d**, and **5a–c** in good to excellent yields, irrespectively of the steric hindrance of the arene and the electron-donating or -withdrawing character of the substituents of the aromatic ring of the sulfonamide partners.

The scope of arene coupling was explored, keeping tosylamide as the nitrogen nucleophile and the promoting system [$Cu(OTf)_2$ /xantphos] in chlorobenzene at 75 °C (Scheme 5). Reacting allyl alcohol and tosylamide with six different arenes bearing electron donor or acceptor heteroatom-based substituents gave the corresponding *N*-tosyl 1-aryl-2-aminopropanes **6–11** in good yields. Worthy of note, this approach is complementary to our previously studied one that used *O*-allyl carbamates as the starting bis-cationic C3

Scheme 3. Synthesis of 1-Mesityl-2-sulfonylamino-propanes 1b–h^{a,b}

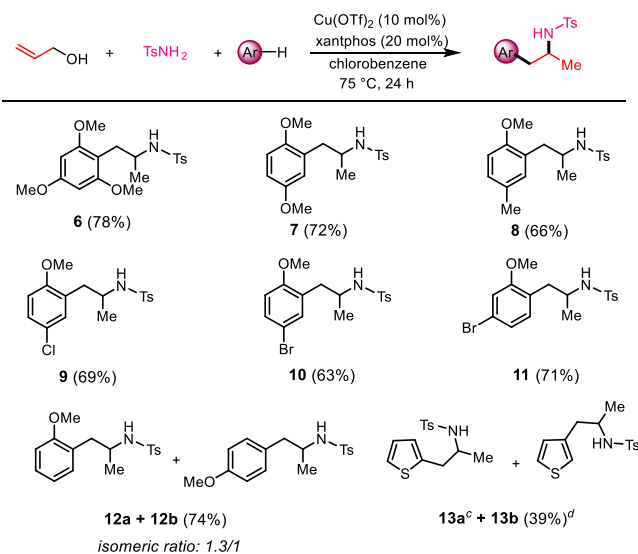
^aReaction conditions: allyl alcohol (1.0 equiv), sulfonamide (2.0 equiv), mesitylene (5.0 equiv), Cu(OTf)₂ (10 mol %), xantphos (20 mol %), and chlorobenzene as the solvent (0.25 M), 75 °C in an oil bath for 24 h. ^bIsolated yields.

Scheme 4. Arylation/Hydroamination with Different Hydrocarbons^{a,b}

^aReaction conditions: allyl alcohol (1.0 equiv), sulfonamide (2.0 equiv), hydrocarbons (5.0 equiv), Cu(OTf)₂ (10 mol %), xantphos (20 mol %), and chlorobenzene as the solvent (0.25 M), 75 °C in an oil bath for 24 h. ^bIsolated yields.

synthetic equivalents.⁵ Indeed, in that case, strongly activated arenes such as 1,4-dimethoxybenzene and 1,3,5-trimethoxybenzene selectively led to 1,2-diarylation products rather than the arylation/hydroamination products 6–7, whereas the reaction carried on with anisole gave a mixture of the two possible regioisomers 12a/12b.

Concerning the use of heteroarenes, furan, indole, and *N*-methylindole furnished only a mixture of degradation products. On the other hand, thiophene afforded the two possible

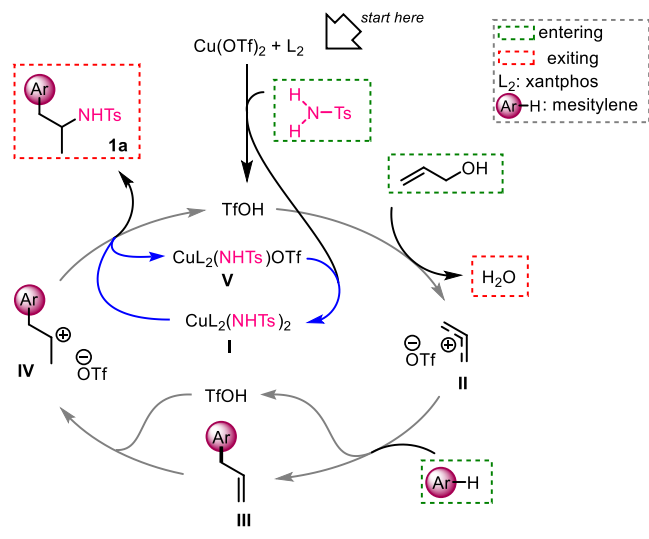
Scheme 5. Arylation/Hydroamination with Different Arenes^{a,b}

^c13a only observed by NMR and not isolated pure. ^d13b isolated with 39% yield. ^aReaction conditions: allyl alcohol (1.0 equiv), tosylamide (2.0 equiv), arene (5.0 equiv), Cu(OTf)₂ (10 mol %), xantphos (20 mol %), and chlorobenzene as the solvent (0.25 M), 75 °C in an oil bath for 24 h. ^bIsolated yields.

products 1-(2-thienyl)-2-tosylaminopropane 13a and 1-(3-thienyl)-2-tosylaminopropane 13b. Only isomer 13b, substituted at the C3 position of the thiophene, was isolated in pure form, whereas isomer 13a was only observed in the crude NMR spectrum.

For the present coupling reaction, we propose the following mechanism (Scheme 6). First, we postulate that the interaction

Scheme 6. Proposed Mechanism for Arylation/Hydroamination of Allyl Alcohol

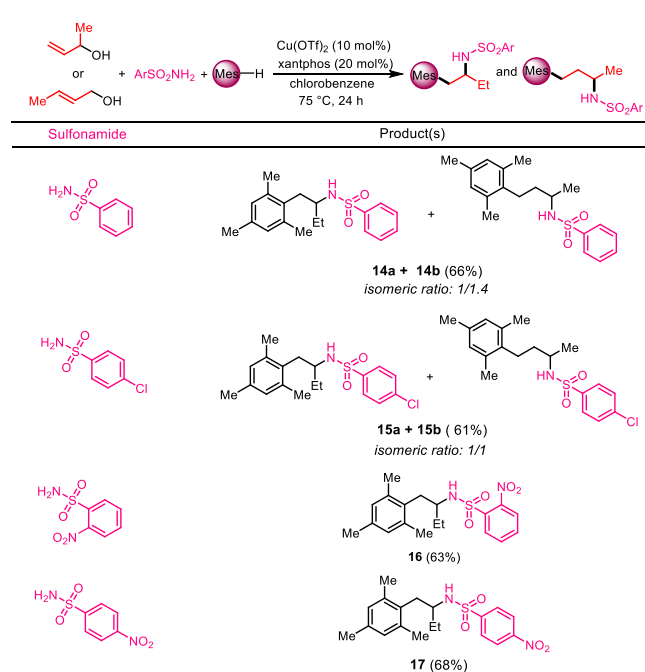


between tosylamide and Cu(OTf)₂ in the presence of the bidentate ligand xantphos generates TfOH and the bis-amido Cu(II) complex CuL₂(NHTs)₂ (I).¹⁴ The following protonation of allyl alcohol generates allyl carbenium ion II accompanied by water release. Subsequent FC allylation of the arene gives the allylated arene III, and its subsequent

Markovnikov protonation by TfOH generates the new carbenium ion **IV**. At this stage, ligand exchange between a sulfonylamino ligand of **I** and triflate anion generates the final product **1a** and the monoamido Cu(II) complex **V**. Finally, the interaction between tosylamide and **V** regenerates **I**.¹⁵ In this mechanism, it is possible to distinguish the double TfOH catalytic cycle (arrows in gray) and the interconnected Cu(II) cycle (arrows in blue).

Finally, the three-component coupling has been tested using substituted allylic alcohols (Scheme 7).¹⁶ On the one hand,

Scheme 7. Variation on the Nature of the Alcohol^{a,b}



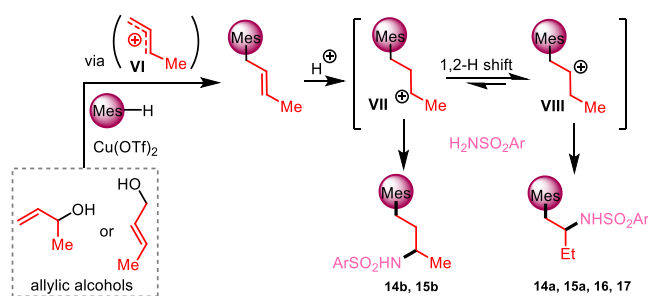
^aReaction conditions: allyl alcohol (1.0 equiv), sulfonamide (2.0 equiv), mesitylene (5.0 equiv), Cu(OTf)₂ (10 mol %), xantphos (20 mol %), and chlorobenzene as the solvent (0.25 M), 75 °C in an oil bath for 24 h. ^bIsolated yields.

reacting crotyl alcohol with mesitylene and benzenesulfonamide or 4-chlorobenzenesulfonamide under the previously optimized conditions gave a 1:1.4 and 1:1 mixture of 3-arylsulfonylamino-4-mesitylbutanes and 2-arylsulfonylamino-4-mesitylbutanes **14a/14b** or **15a/15b** in 66 and 61% isolated yields, respectively. On the other hand, using 2-nitrobenzenesulfonamide or 4-nitrobenzenesulfonamide as the nitrogen nucleophile gave exclusively the 3-arylsulfonylamino-4-mesitylbutanes **16** and **17** in 63 and 68% yields, respectively. Repeating the same four couplings as above using 3-buten-2-ol instead of crotyl alcohol gave precisely the same results. Thus, the two isomeric allylic alcohols can act in this three-component coupling as butane-1,2-dylium or butane-1,3-dylium C4 synthons. Conversely, treatment of α,α - and γ,γ -dimethyl-substituted allyl alcohols afforded only diarylated derivatives with indane structures.^{5,17}

The above outcome can be interpreted as follows. Protonation of crotyl alcohol or 3-buten-2-ol generates the common allylic carbenium ion **VI** that is intercepted by the arene to give a crotylated arene. Further protonation of this latter at position 2 or 3 of the chain generates the transient carbenium ions **VII** and **VIII**, which can in turn be trapped by

the sulfonamides to give the two regioisomeric final products (Scheme 8).¹⁸

Scheme 8. Key Intermediates for Arylation/Hydroamination of Butenol Substrates



CONCLUSIONS

In conclusion, we have shown that simple allylic alcohols are ideal C3 (or higher) bis-cationic alkane-1,2-dylium synthons in [FC allylation/hydroamination] cascades. This copper-catalyzed three-component reaction discloses a novel, straightforward, and general preparation of the pharmacologically relevant class of 1-aryl-2-aminopropanes. Future studies will be addressed to test new nucleophiles and intramolecular variants.

EXPERIMENTAL SECTION

General Information. All available chemicals and solvents were purchased from commercial sources and were used without any further purification. Thin-layer chromatography (TLC) was performed using 0.25 mm silica gel pre-coated plates Si 60-F254 (Merck, Darmstadt, Germany) visualized by UV-254 light and cerium ammonium molybdate (CAM) staining. Purification by flash column chromatography (FCC) was conducted by using silica gel Si 60, 230–400 mesh, 0.040–0.063 mm (Merck). Melting points were determined on a Stuart Scientific SMP3 and are corrected. ¹H and ¹³C NMR spectra were recorded on a Bruker Avance 400 (400 and 101 MHz, respectively); chemical shifts are indicated in parts per million downfield from SiMe₄, using the residual proton (CHCl₃ = 7.27 ppm) and carbon (CDCl₃ = 77.0 ppm) solvent resonances as an internal reference. Coupling constant values *J* are given in Hz. High-resolution mass spectra (HRMS) were recorded using a mass spectrometer MicroTOF from Bruker with an electron spray ion source (ESI) and a TOF detector or using a mass spectrometer from Thermo Fisher Scientific with an electron spray ion source (ESI) and a LTQ Orbitrap as a detector. FTIR spectra were recorded on a Tensor 27 (ATR Diamond) Bruker infrared spectrophotometer and are reported in frequency of absorption (cm⁻¹).

Safety Note. TfOH is a strong protic acid and corrosive; therefore, it requires careful handling. All reactions should be carried out with safety precautions in a ventilated hood using protective clothing.

General Procedure for the Synthesis of Arylated/Hydroaminated Products with Allyl Alcohol. In a sealed tube, after 15 min, the allyl alcohol (1.0 mmol, 58 mg), hydrocarbons (5.0 mmol), and sulfonamide (2.0 mmol) were added to a solution of Cu(OTf)₂ (10 mol %, 36.2 mg) and xantphos (20 mol %, 11.6 mg) in chlorobenzene (0.25 M). The resulted solution was magnetically stirred and heated at 75 °C in an oil bath for 24 h. The reaction mixture was washed with brine (3 × 5 mL) and the organic layer was extracted with AcOEt (2 × 5 mL), dried over MgSO₄, and filtered. The solvent was evaporated under reduced pressure. The residue was purified by FCC. Starting from the aromatic source and appropriate sulfonamide, yield, physical, spectroscopic, and analytical data of compounds **1a–f**, **3a–f**, **4a–d**, **5a–b**, and **6–11** are as follows.

1-(2,4,6-Trimethylphenyl)-2-tosylamino-propane (1a). Mesitylene (0.69 mL); tosylamide (342.4 mg); FCC–AcOEt/hexane (1:1). **1a** (268.2 mg, 81%); light brown oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 7.58 (d, 2H, $J = 8.3$ Hz), 7.19 (d, 2H, $J = 7.9$ Hz), 6.74 (s, 2H), 4.43 (d, 1H, $J = 7.2$ Hz), 3.49–3.42 (m, 1H), 2.79 (dd, 1H, $J = 14.0, 7.2$ Hz), 2.64 (dd, 1H, $J = 14.0, 8.1$ Hz), 2.41 (s, 3H), 2.23 (s, 3H), 2.14 (s, 6H), 1.14 (d, 3H, $J = 6.5$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 143.1, 137.5, 136.6, 135.8, 131.5, 129.5, 129.2, 127.0, 49.9, 36.9, 21.5, 21.4, 20.8, 20.3. The characterization of product **1a** is consistent with that reported in the literature.⁵

1-(2,4,6-Trimethylphenyl)-2-(*p*-nosylamino)-propane (1b). Mesitylene (0.69 mL); *p*-nosylamide (404.4 mg); FCC–AcOEt/hexane (1:4), R_f : 0.29. **1b** (329.5 mg, 91%); light brown oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 8.12 (d, 2H, $J = 8.7$ Hz), 7.69 (d, 2H, $J = 8.7$ Hz), 6.64 (s, 2H), 4.49 (d, 1H, $J = 7.8$ Hz), 3.62–3.55 (m, 1H), 2.73 (dd, 1H, $J = 14.2, 9.2$ Hz), 2.65 (dd, 1H, $J = 14.4, 5.8$ Hz), 2.19 (s, 3H), 2.13 (s, 6H), 1.32 (d, 3H, $J = 6.4$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 149.5, 146.0, 136.4, 136.2, 131.1, 129.3, 127.7, 123.8, 50.9, 36.5, 23.1, 20.6, 20.2; IR ν_{max} 2918, 1342, 1158 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{18}\text{H}_{21}\text{N}_2\text{O}_4\text{S}$ $[\text{M} - \text{H}]^-$: 361.1228; found: 361.1216.

1-(2,4,6-Trimethylphenyl)-2-(4-chlorobenzenesulfonamido)-propane (1c). Mesitylene (0.69 mL); 4-chlorobenzenesulfonamide (383.3 mg); FCC–AcOEt/hexane (3:7), R_f : 0.33. **1c** (242.3 mg, 69%); light brown oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 7.53 (d, 2H, $J = 8.4$ Hz), 7.30 (d, 2H, $J = 8.4$ Hz), 6.71 (s, 2H), 4.49 (d, 1H, $J = 7.3$ Hz), 3.53–3.44 (m, 1H), 2.76 (dd, 1H, $J = 14.1, 8.2$ Hz), 2.64 (dd, 1H, $J = 14.2, 6.9$ Hz), 2.25 (s, 3H), 2.14 (s, 6H), 1.23 (d, 3H, $J = 6.4$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 138.8, 138.7, 136.3, 136.1, 131.2, 129.3, 128.9, 128.2, 50.3, 36.6, 22.4, 20.8, 20.3; IR ν_{max} 2923, 1321, 1163 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{18}\text{H}_{21}\text{ClNO}_2\text{S}$ $[\text{M} - \text{H}]^-$: 350.0987; found: 350.0977.

1-(2,4,6-Trimethylphenyl)-2-(4-trifluoromethylbenzenesulfonamido)-propane (1d). Mesitylene (0.69 mL); 4-(trifluoromethyl)benzenesulfonamide (450.4 mg); FCC–AcOEt/hexane (1:9), R_f : 0.31. **1d** (277.3 mg, 72%); colorless oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 7.69 (d, 2H, $J = 8.2$ Hz), 7.58 (d, 2H, $J = 8.3$ Hz), 6.68 (s, 2H), 4.56 (d, 1H, $J = 7.4$ Hz), 3.55–3.48 (m, 1H), 2.75 (dd, 1H, $J = 14.2, 8.6$ Hz), 2.65 (dd, 1H, $J = 14.2, 6.6$ Hz), 2.23 (s, 3H), 2.12 (s, 6H), 1.27 (d, 3H, $J = 6.0$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 143.8, 136.3, 136.1, 133.9 (q, $\text{JC}-\text{F} = 33.0$ Hz), 131.1, 129.3, 127.2, 125.8 (q, $\text{JC}-\text{F} = 3.7$ Hz), 124.7 (q, $\text{JC}-\text{F} = 253.6$ Hz), 50.4, 36.5, 22.6, 20.6, 20.2; IR ν_{max} 2922, 1339, 1125, 1018 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{19}\text{H}_{21}\text{F}_3\text{NO}_2\text{S}$ $[\text{M} - \text{H}]^-$: 384.1251; found: 384.1240.

1-(2,4,6-Trimethylphenyl)-2-(2-methylbenzenesulfonamido)-propane (1e). Mesitylene (0.69 mL); 2-methylbenzenesulfonamide (342.4 mg); FCC–AcOEt/hexane (1:4), R_f : 0.40. **1e** (291.4 mg, 88%); light brown oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 7.85 (d, 1H, $J = 7.8$ Hz), 7.33 (t, 1H, $J = 7.4$ Hz), 7.17 (t, 1H, $J = 6.6$ Hz), 7.10 (d, 1H, $J = 7.6$ Hz), 6.64 (s, 2H), 4.36 (d, 1H, $J = 7.1$ Hz), 3.36–3.29 (m, 1H), 2.69 (dd, 1H, $J = 14.0, 7.4$ Hz), 2.56 (dd, 1H, $J = 14.0, 8.3$ Hz), 2.29 (s, 3H), 2.14 (s, 3H), 2.01 (s, 6H), 1.08 (d, 3H, $J = 6.4$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 137.9, 137.2, 136.6, 135.9, 132.6, 132.5, 131.2, 129.6, 129.3, 125.9, 49.7, 36.7, 21.8, 20.8, 20.2, 20.0; IR ν_{max} 2917, 1299, 1157 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{19}\text{H}_{24}\text{NO}_2\text{S}$ $[\text{M} - \text{H}]^-$: 330.1533; found: 330.1532.

1-(2,4,6-Trimethylphenyl)-2-(*o*-nosylamino)-propane (1f). Mesitylene (0.69 mL); *o*-nosylamide (404.4 mg); FCC–AcOEt/hexane (1:4). **1f** (336.8 mg, 93%); light brown oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 7.98 (d, 1H, $J = 7.4$ Hz), 7.79 (d, 1H, $J = 7.6$ Hz), 7.68–7.61 (m, 2H), 6.62 (s, 2H), 5.31 (t, 1H, $J = 3.4$ Hz), 3.85–3.78 (m, 1H), 2.85 (dd, 1H, $J = 14.2, 8.0$ Hz), 2.72 (dd, 1H, $J = 14.2, 7.5$ Hz) 2.19 (s, 6H), 2.16 (s, 3H), 1.26 (d, 3H, $J = 7.5$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 136.5, 136.4, 135.7, 134.8, 132.8, 131.1, 130.4, 129.2, 125.44, 125.43, 51.2, 36.6, 22.4, 20.7, 20.3. The characterization of product **1f** is consistent with that reported in the literature.⁵

1-(2,4,6-Trimethylphenyl)-2-(3-methoxybenzenesulfonamido)-propane (1g). Mesitylene (0.69 mL); 3-methoxybenzenesulfonamide (374.4 mg); FCC–AcOEt/hexane (2:3), R_f : 0.37. **1g** (215.2 mg, 62%); yellow oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 7.31–7.23 (m, 2H),

7.19 (s, 1H), 6.99 (d, 1H, $J = 7.3$ Hz), 6.67 (s, 2H), 4.81 (d, 1H, $J = 6.9$ Hz), 3.75 (s, 3H), 3.46–3.38 (m, 1H), 2.78 (dd, 1H, $J = 13.9, 7.0$ Hz), 2.59 (dd, 1H, $J = 13.9, 8.3$ Hz), 2.16 (s, 3H), 2.09 (s, 6H), 1.08 (d, 3H, $J = 6.5$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 159.8, 141.6, 136.5, 135.8, 131.3, 129.9, 129.3, 119.2, 118.9, 111.4, 55.5, 50.0, 36.8, 21.6, 20.8, 20.3; IR ν_{max} 2968, 1309, 1155 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{19}\text{H}_{24}\text{NO}_3\text{S}$ $[\text{M} - \text{H}]^-$: 346.1482; found: 346.1468.

1-(2,4,6-Trimethylphenyl)-2-(benzenesulfonamido)-propane (1h). Mesitylene (0.69 mL); benzenesulfonamide (314.4 mg); FCC–AcOEt/hexane (1.5:8.5), R_f : 0.36. **1h** (263.2 mg, 83%); colorless oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 7.68 (d, 2H, $J = 7.5$ Hz), 7.52 (t, 1H, $J = 7.3$ Hz), 7.40 (t, 2H, $J = 7.7$ Hz), 6.74 (s, 2H), 4.42 (d, 1H, $J = 6.8$ Hz), 3.51–3.44 (m, 1H), 2.79 (dd, 1H, $J = 13.9, 7.3$ Hz), 2.64 (dd, 1H, $J = 13.9, 7.9$ Hz), 2.23 (s, 3H), 2.14 (s, 6H), 1.16 (d, 3H, $J = 6.5$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 140.3, 136.5, 135.9, 132.3, 131.2, 129.3, 128.8, 126.9, 49.9, 36.8, 21.8, 20.8, 20.2; IR ν_{max} 2936, 1326, 1158 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{18}\text{H}_{22}\text{NO}_2\text{S}$ $[\text{M} - \text{H}]^-$: 316.1377; found: 316.1376.

1-(2,3,5,6-Tetramethylphenyl)-2-tosylamino-propane (3a). 1,2,4,5-Tetramethylbenzene (671.1 mg), tosylamide (342.4 mg); FCC–AcOEt/hexane (4:1). **3a** (286.5 mg, 83%); light brown oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 7.49 (d, 2H, $J = 8.3$ Hz), 7.15 (d, 2H, $J = 8.2$ Hz), 6.82 (s, 1H), 4.51 (d, 1H, $J = 6.7$ Hz), 3.45–3.38 (m, 1H), 2.90 (dd, 1H, $J = 14.3, 7.8$ Hz), 2.77 (dd, 1H, $J = 14.3, 7.5$ Hz), 2.41 (s, 3H), 2.16 (s, 6H), 2.04 (s, 6H), 1.19 (d, 3H, $J = 6.5$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 142.9, 137.2, 134.1, 133.9, 132.5, 130.2, 129.3, 126.9, 50.3, 37.2, 21.9, 21.5, 20.7, 16.1. The characterization of product **3a** is consistent with that reported in the literature.⁵

1-(2,3,5,6-Tetramethylphenyl)-2-(benzenesulfonamido)-propane (3b). 1,2,4,5-Tetramethylbenzene (671.1 mg), benzenesulfonamide (314.4 mg); FCC–DCM/MeOH (9.9:0.1), R_f : 0.35. **3b** (281.5 mg, 85%); colorless oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 7.60 (d, 2H, $J = 8.0$ Hz), 7.49 (t, 1H, $J = 7.4$ Hz), 7.37 (t, 2H, $J = 7.9$ Hz), 6.82 (s, 1H), 4.44 (d, 1H, $J = 6.7$ Hz), 3.45–3.38 (m, 1H), 2.90 (dd, 1H, $J = 14.3, 7.8$ Hz), 2.77 (dd, 1H, $J = 14.3, 7.5$ Hz), 2.16 (s, 6H), 2.19 (d, 3H, $J = 6.5$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 140.1, 134.0, 133.9, 132.5, 132.2, 130.4, 128.8, 126.9, 50.3, 37.2, 21.9, 20.7, 16.1; IR ν_{max} 2920, 1379, 1135 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{19}\text{H}_{25}\text{NO}_2\text{S}$ $[\text{M} - \text{H}]^-$: 330.1533; found: 330.1531.

1-(2,3,5,6-Tetramethylphenyl)-2-(4-chlorobenzenesulfonamido)-propane (3c). 1,2,4,5-Tetramethylbenzene (671.1 mg), 4-chlorobenzenesulfonamide (383.3 mg); FCC–AcOEt/hexane (1:4), R_f : 0.34. **3c** (259.2 mg, 71%); light brown oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 7.41 (d, 2H, $J = 8.5$ Hz), 7.25 (d, 2H, $J = 8.6$ Hz), 6.82 (s, 1H), 4.32 (d, 1H, $J = 7.2$ Hz), 3.46–3.39 (m, 1H), 2.86 (dd, 1H, $J = 14.5, 8.9$ Hz), 2.75 (dd, 1H, $J = 14.5, 6.2$ Hz), 2.15 (s, 6H), 2.02 (s, 6H), 1.29 (d, 3H, $J = 6.4$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 138.6, 138.5, 134.1, 133.8, 132.3, 130.3, 128.8, 128.1, 50.7, 36.9, 22.7, 20.6, 16.1; IR ν_{max} 2917, 1381, 1135 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{19}\text{H}_{23}\text{ClNO}_2\text{S}$ $[\text{M} - \text{H}]^-$: 364.1144; found: 364.1129.

1-(2,3,5,6-Tetramethylphenyl)-2-(*p*-nosylamino)-propane (3d). 1,2,4,5-Tetramethylbenzene (671.1 mg), 4-nitrobenzenesulfonamide (404.4 mg); FCC–AcOEt/hexane (1:4), R_f : 0.27. **3d** (346.1 mg, 92%); light brown oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 8.08 (d, 2H, $J = 8.7$ Hz), 7.59 (d, 2H, $J = 8.7$ Hz), 6.73 (s, 1H), 4.53 (d, 1H, $J = 7.8$ Hz), 3.58–3.49 (m, 1H), 2.85 (dd, 1H, $J = 14.5, 9.6$ Hz), 2.74 (dd, 1H, $J = 14.8, 5.5$ Hz), 2.09 (s, 6H), 2.02 (s, 6H), 1.37 (d, 3H, $J = 6.4$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 149.5, 145.8, 134.1, 133.9, 132.2, 130.3, 127.5, 123.6, 51.4, 36.9, 23.3, 20.5, 16.1; IR ν_{max} 2920, 1345, 1160 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{19}\text{H}_{23}\text{N}_2\text{O}_4\text{S}$ $[\text{M} - \text{H}]^-$: 375.1384; found: 375.1368.

1-(2,3,5,6-Tetramethylphenyl)-2-(2-methylbenzenesulfonamido)-propane (3e). 1,2,4,5-Tetramethylbenzene (671.1 mg), 2-methylbenzenesulfonamide (342.4 mg); FCC–AcOEt/hexane (1:4), R_f : 0.34. **3e** (255.4 mg, 74%); yellow oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 7.90 (d, 1H, $J = 7.8$ Hz), 7.40 (t, 1H, $J = 9.0$ Hz), 7.25–7.23 (m, 1H), 7.14 (d, 1H, $J = 7.4$ Hz), 6.82 (s, 1H), 4.36 (d, 1H, $J = 6.2$ Hz), 3.39–3.22 (m, 1H), 2.88 (dd, 1H, $J = 14.3, 8.2$ Hz), 2.76 (dd, 1H, $J = 14.3, 7.4$ Hz), 2.26 (s, 3H), 2.15 (s, 6H), 1.97 (s, 6H), 1.24

(d, 3H, $J = 6.5$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 137.5, 137.1, 134.1, 133.8, 132.6, 132.5, 132.4, 130.4, 129.7, 125.8, 50.0, 37.1, 22.1, 20.6, 19.8, 16.0; IR ν_{max} 2918, 1315, 1125 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{20}\text{H}_{26}\text{NO}_2\text{S}$ [$\text{M} - \text{H}$] $^-$: 344.1690; found: 344.1686.

1-(2,3,5,6-Tetramethylphenyl)-2-(*o*-nosylamino)-propane (3f). 1,2,4,5-Tetramethylbenzene (671.1 mg), 2-nitrobenzenesulfonamide (404.4 mg); FCC–AcOEt/hexane (4:1), R_f : 0.37. **3f** (327.2 mg, 87%); light brown oil. ^1H NMR (CDCl_3 , 400 MHz) δ 7.95–7.92 (m, 1H), 7.79–7.76 (m, 1H), 7.64–7.62 (m, 2H), 6.66 (s, 1H), 5.33 (d, 1H, $J = 6.4$ Hz), 3.78–3.71 (m, 1H), 2.97 (dd, 1H, $J = 14.6, 8.6$ Hz), 2.84 (dd, 1H, $J = 14.5, 6.9$ Hz), 2.08 (s, 6H), 2.07 (s, 6H), 1.31 (d, 3H, $J = 6.5$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 146.9, 134.6, 133.9, 133.8, 132.9, 132.6, 132.5, 130.34, 130.32, 125.5, 51.9, 36.9, 22.8, 20.6, 16.2; IR ν_{max} 2920, 1344, 1163 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{19}\text{H}_{23}\text{N}_2\text{O}_4\text{S}$ [$\text{M} - \text{H}$] $^-$: 375.1384; found: 375.1375.

1-(2,3,4,5,6-Pentamethylphenyl)-2-tosylamino-propane (4a). 1,2,3,4,5-Pentamethylbenzene (741.2 mg), tosylamide (342.4 mg); FCC–DCM, R_f : 0.41. **4a** (305.3 mg, 85%); white solid, mp 169–170 $^\circ\text{C}$. ^1H NMR (CDCl_3 , 400 MHz) δ 7.46 (d, 2H, $J = 8.2$ Hz), 7.12 (d, 2H, $J = 8.0$ Hz), 4.29 (d, 1H, $J = 6.4$ Hz), 3.41–3.34 (m, 1H), 2.91 (dd, 1H, $J = 14.5, 7.9$ Hz), 2.78 (dd, 1H, $J = 14.5, 7.3$ Hz), 2.40 (s, 3H), 2.22 (s, 3H), 2.14 (s, 6H), 2.08 (s, 6H), 1.19 (d, 3H, $J = 6.4$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 142.8, 137.1, 133.3, 132.8, 132.1, 131.2, 129.2, 126.9, 50.5, 37.5, 21.9, 21.5, 17.1, 16.9, 16.8; IR ν_{max} 2919, 1319, 1135 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{21}\text{H}_{28}\text{NO}_2\text{S}$ [$\text{M} - \text{H}$] $^-$: 358.1846; found: 358.1835.

1-(2,3,4,5,6-Pentamethylphenyl)-2-(*p*-nosylamino)-propane (4b). 1,2,3,4,5-Pentamethylbenzene (741.2 mg), 4-nitrobenzenesulfonamide (404.4 mg); FCC–AcOEt/hexane (2:3), R_f : 0.35. **4b** (343.3 mg, 88%); orange solid; mp 183–185 $^\circ\text{C}$. ^1H NMR (CDCl_3 , 400 MHz) δ 8.03 (d, 2H, $J = 8.7$ Hz), 7.57 (d, 2H, $J = 8.7$ Hz), 4.46 (d, 1H, $J = 8.2$ Hz), 3.55–3.48 (m, 1H), 2.86 (dd, 1H, $J = 14.8, 9.8$ Hz), 2.76 (dd, 1H, $J = 14.9, 5.2$ Hz), 2.15 (s, 3H), 2.07 (s, 12H), 1.38 (d, 3H, $J = 6.4$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 149.2, 145.8, 133.8, 132.9, 131.7, 131.1, 127.6, 123.4, 51.7, 37.1, 23.4, 17.1, 16.8, 16.7; IR ν_{max} 2920, 1349, 1165 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{20}\text{H}_{25}\text{N}_2\text{O}_4\text{S}$ [$\text{M} - \text{H}$] $^-$: 389.1541; found: 389.1530.

1-(2,3,4,5,6-Pentamethylphenyl)-2-(2-methylbenzenesulfonamido)-propane (4c). 1,2,3,4,5-Pentamethylbenzene (741.2 mg), 2-methylbenzenesulfonamide (342.4 mg); FCC–AcOEt/hexane (2:3), R_f : 0.35. **4c** (276.6 mg, 77%); brown solid; mp 115–117 $^\circ\text{C}$. ^1H NMR (CDCl_3 , 400 MHz) δ 7.92 (d, 1H, $J = 7.8$ Hz), 7.42 (t, 1H, $J = 7.4$ Hz), 7.26 (t, 1H, $J = 7.6$ Hz), 7.12 (d, 1H, $J = 7.5$ Hz), 4.50 (d, 1H, $J = 6.4$ Hz), 3.42–3.32 (m, 1H), 2.92 (dd, 1H, $J = 14.5, 8.2$ Hz), 2.81 (dd, 1H, $J = 14.5, 7.3$ Hz), 2.28 (s, 3H), 2.23 (s, 3H), 2.15 (s, 6H), 2.06 (s, 6H), 1.27 (d, 3H, $J = 6.4$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 137.6, 137.1, 133.4, 132.8, 132.4, 132.2, 132.1, 131.2, 129.7, 125.7, 50.4, 37.4, 22.2, 19.8, 17.0, 16.94, 16.91; IR ν_{max} 2932, 1347, 1126 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{21}\text{H}_{28}\text{NO}_2\text{S}$ [$\text{M} - \text{H}$] $^-$: 358.1846; found: 358.1835.

1-(2,3,4,5,6-Pentamethylphenyl)-2-(*o*-nosylamino)-propane (4d). 1,2,3,4,5-Pentamethylbenzene (741.2 mg), 2-nitrobenzenesulfonamide (404.4 mg); FCC–AcOEt/hexane (2:3), R_f : 0.33. **4d** (347.5 mg, 89%); yellow solid; mp 130–132 $^\circ\text{C}$. ^1H NMR (CDCl_3 , 400 MHz) δ 7.92 (d, 1H, $J = 9.2$ Hz), 7.74 (d, 1H, $J = 9.4$ Hz), 7.65–7.57 (m, 2H), 5.35 (d, 1H, $J = 6.3$ Hz), 3.77–3.70 (m, 1H), 2.98 (dd, 1H, $J = 14.8, 8.9$ Hz), 2.86 (dd, 1H, $J = 14.8, 6.6$ Hz), 2.13 (s, 6H), 2.11 (s, 3H), 2.04 (s, 6H), 1.33 (d, 3H, $J = 6.5$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 146.7, 134.6, 133.1, 132.7, 132.6, 132.5, 132.1, 131.1, 130.4, 125.3, 52.0, 37.1, 22.9, 17.2, 16.9, 16.8; IR ν_{max} 2922, 1346, 1128 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{20}\text{H}_{25}\text{N}_2\text{O}_4\text{S}$ [$\text{M} - \text{H}$] $^-$: 389.1541; found: 389.1530.

1-(2,5-Dimethylphenyl)-2-tosylamino-propane (5a). *p*-Xylene (0.62 mL); tosylamide (342.4 mg); FCC–AcOEt/hexane (2:3). **5a** (231.5 mg, 73%); light brown oil. ^1H NMR (CDCl_3 , 400 MHz) δ 7.59 (d, 2H, $J = 8.2$ Hz), 7.20 (d, 2H, $J = 8.1$ Hz), 6.94–6.89 (m, 2H), 6.76 (s, 1H), 4.72 (d, 1H, $J = 6.7$ Hz), 3.48–3.41 (m, 1H), 2.73 (dd, 1H, $J = 13.7, 6.9$ Hz), 2.59 (dd, 1H, $J = 13.7, 7.4$ Hz), 2.41 (s, 3H), 2.24 (s, 3H), 2.09 (s, 3H), 1.15 (d, 3H, $J = 6.4$ Hz); $^{13}\text{C}\{^1\text{H}\}$

NMR (CDCl_3 , 101 MHz) δ 143.0, 137.5, 135.4, 135.3, 133.2, 130.9, 130.5, 129.5, 127.5, 126.9, 50.1, 41.1, 21.7, 21.5, 20.9, 18.8. The characterization of product **5a** is consistent with that reported in the literature.⁵

1-(2,5-Dimethylphenyl)-2-(*p*-nosylamino)-propane (5b). *p*-Xylene (0.62 mL); 4-nitrobenzenesulfonamide (404.4 mg); FCC–AcOEt/hexane (1:2), R_f : 0.31. **5b** (261.1 mg, 75%); light brown oil. ^1H NMR (CDCl_3 , 400 MHz) δ 8.13 (d, 2H, $J = 8.8$ Hz), 7.69 (d, 2H, $J = 8.8$ Hz), 6.89–6.84 (m, 2H), 6.69 (s, 1H), 4.49 (d, 1H, $J = 7.4$ Hz), 3.58–3.48 (m, 1H), 2.72 (dd, 1H, $J = 14.0, 5.5$ Hz), 2.58 (dd, 1H, $J = 14.0, 8.9$ Hz), 2.19 (s, 3H), 2.09 (s, 3H), 1.31 (d, 3H, $J = 6.4$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 149.6, 149.1, 146.1, 135.6, 135.1, 130.8, 130.6, 127.8 (2CH), 123.9, 51.1, 40.9, 23.0, 20.8, 18.8; IR ν_{max} 2921, 1377, 1161 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{17}\text{H}_{19}\text{N}_2\text{O}_4\text{S}$ [$\text{M} - \text{H}$] $^-$: 347.1071; found: 347.1055.

1-(2,5-Dimethylphenyl)-2-(4-chlorobenzenesulfonamido)-propane (5c). *p*-Xylene (0.62 mL); 4-chlorobenzenesulfonamide (383.3 mg); FCC–DCM/MeOH (9.9:0.1), R_f : 0.43. **5c** (229.2 mg, 68%); light brown oil. ^1H NMR (CDCl_3 , 400 MHz) δ 7.54 (d, 2H, $J = 8.5$ Hz), 7.32 (d, 2H, $J = 8.6$ Hz), 6.92 (s, 2H), 6.73 (s, 1H), 4.68 (d, 1H, $J = 6.9$ Hz), 3.49–3.42 (m, 1H), 2.64 (d, 2H, $J = 7.2$ Hz), 2.23 (s, 3H), 2.10 (s, 3H), 1.23 (d, 3H, $J = 6.4$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 138.9, 138.7, 135.5, 135.3, 132.9, 130.8, 130.6, 129.1, 128.2, 127.6, 50.5, 41.0, 22.3, 20.8, 18.8; IR ν_{max} 2924, 1322, 1159 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{17}\text{H}_{19}\text{ClNO}_2\text{S}$ [$\text{M} - \text{H}$] $^-$: 336.0831; found: 336.0828.

1-(2,4,6-Trimethoxyphenyl)-2-tosylamino-propane (6). 1,3,5-Trimethoxybenzene (890.9 mg); tosylamide (342.4 mg); FCC–AcOEt/hexane (2:3), R_f : 0.34. **6** (295.7 mg, 78%); colorless oil. ^1H NMR (CDCl_3 , 400 MHz) δ 7.34 (d, 2H, $J = 8.1$ Hz), 7.00 (d, 2H, $J = 8.0$ Hz), 5.93 (s, 2H), 5.09 (d, 1H, $J = 5.4$ Hz), 3.80 (s, 3H), 3.69 (s, 6H), 3.32–3.26 (m, 1H), 2.61–2.49 (m, 2H), 2.37 (s, 3H), 1.30 (d, 3H, $J = 6.4$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 160.0, 158.4, 141.9, 137.1, 128.9, 126.6, 106.6, 90.4, 55.5, 55.2, 50.8, 29.4, 23.5, 21.4; IR ν_{max} 2920, 1379, 1207, 1160 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{19}\text{H}_{24}\text{NO}_5\text{S}$ [$\text{M} - \text{H}$] $^-$: 378.1381; found: 378.1359.

1-(2,5-Dimethoxyphenyl)-2-tosylamino-propane (7). 1,4-Dimethoxybenzene (690.8 mg); tosylamide (342.4 mg); FCC–AcOEt/hexane (2:3), R_f : 0.32. **7** (251.4 mg, 72%); colorless oil. ^1H NMR (CDCl_3 , 400 MHz) δ 7.43 (d, 2H, $J = 8.1$ Hz), 7.08 (d, 2H, $J = 8.9$ Hz), 6.68 (s, 2H), 6.41 (s, 1H), 5.06 (d, 1H, $J = 5.4$ Hz), 3.74 (s, 3H), 3.69 (s, 3H), 3.45–3.36 (m, 1H), 2.73 (dd, 1H, $J = 13.6, 9.0$ Hz), 2.49 (dd, 1H, $J = 13.6, 4.9$ Hz), 2.37 (s, 3H), 1.25 (d, 3H, $J = 6.6$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 153.7, 151.3, 142.5, 137.1, 129.2, 127.0, 126.8, 116.7, 112.4, 111.5, 55.9, 55.5, 51.2, 37.5, 22.8, 21.4; IR ν_{max} 2929, 1223, 1155 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{18}\text{H}_{22}\text{NO}_4\text{S}$ [$\text{M} - \text{H}$] $^-$: 348.1275; found: 348.1273.

1-(5-Methylanisole)-2-tosylamino-propane (8). 4-Methylanisole (0.63 mL); tosylamide (342.4 mg); FCC–DCM, R_f : 0.31. **8** (219.9 mg, 66%); yellow oil. ^1H NMR (CDCl_3 , 400 MHz) δ 7.42 (d, 2H, $J = 8.2$ Hz), 7.08 (d, 2H, $J = 8.1$ Hz), 6.94 (d, 1H, $J = 8.3$ Hz), 6.66–6.63 (m, 2H), 4.99 (d, 1H, $J = 5.3$ Hz), 3.74 (s, 3H), 3.45–3.38 (m, 1H), 2.69 (dd, 1H, $J = 13.6, 8.9$ Hz), 2.51 (dd, 1H, $J = 13.6, 4.9$ Hz), 2.38 (s, 3H), 2.18 (s, 3H), 1.25 (d, 3H, $J = 6.6$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 169.4, 142.4, 137.2, 131.8, 130.0, 129.2, 128.2, 126.8, 125.7, 110.4, 55.4, 51.2, 37.3, 27.1, 22.8, 21.4. The characterization of product **8** is consistent with that reported in the literature.⁵

1-(5-Chloroanisole)-2-tosylamino-propane (9). 4-Chloroanisole (0.63 mL); tosylamide (342.4 mg); FCC–AcOEt/hexane (2:3), R_f : 0.33. **9** (243.6 mg, 69%); light yellow oil. ^1H NMR (CDCl_3 , 400 MHz) δ 7.44 (d, 2H, $J = 8.0$ Hz), 7.11 (d, 1H, $J = 7.9$ Hz), 7.09–7.06 (m, 1H), 6.80 (d, 1H, $J = 2.2$ Hz), 6.65 (d, 1H, $J = 8.7$ Hz), 4.79 (d, 1H, $J = 5.9$ Hz), 3.77 (s, 3H), 3.45–3.42 (m, 1H), 2.71 (dd, 1H, $J = 13.5, 9.3$ Hz), 2.49 (dd, 1H, $J = 13.7, 5.0$ Hz), 2.39 (s, 3H), 1.26 (d, 3H, $J = 5.9$ Hz); ^{13}C NMR (CDCl_3 , 101 MHz) δ 155.7, 144.3, 142.8, 130.7, 129.3, 127.9, 127.5, 126.7, 125.7, 111.6, 55.7, 51.1, 37.1, 22.9, 21.5; IR ν_{max} 2918, 1326, 1157 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{17}\text{H}_{19}\text{ClNO}_3\text{S}$ [$\text{M} - \text{H}$] $^-$: 352.0780; found: 352.0778.

1-(5-Bromoanisole)-2-tosylamino-propane (10). 4-Bromoanisole (0.63 mL); tosylamide (342.4 mg); FCC–AcOEt/hexane (2:3). **10** (250.1 mg, 63%); yellow oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 7.43 (d, 2H, $J = 8.2$ Hz), 7.23–7.20 (m, 1H), 7.11 (d, 2H, $J = 8.1$ Hz), 6.95 (d, 1H, $J = 2.3$ Hz), 6.59 (d, 1H, $J = 8.7$ Hz), 4.79 (d, 1H, $J = 6.2$ Hz), 3.77 (s, 3H), 3.47–3.39 (m, 1H), 2.71 (dd, 1H, $J = 13.6, 10.7$ Hz), 2.49 (dd, 1H, $J = 13.6, 4.9$ Hz), 2.40 (s, 3H), 1.26 (d, 3H, $J = 3.6$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 156.2, 142.8, 136.9, 133.5, 130.5, 129.4, 128.4, 126.7, 113.1, 112.1, 55.6, 51.2, 37.1, 23.0, 21.5. The characterization of product **10** is consistent with that reported in the literature.⁵

1-(5-Bromoanisole)-2-tosylamino-propane (11). 3-Bromoanisole (0.63 mL); tosylamide (342.4 mg); FCC–AcOEt/hexane (1:4), R_f : 0.29. **11** (281.9 mg, 71%); light brown oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 7.54 (d, 2H, $J = 8.3$ Hz), 7.15 (d, 2H, $J = 8.3$ Hz), 6.93 (d, 2H, $J = 7.8$ Hz), 6.67 (dd, 1H, $J = 8.4, 2.6$ Hz), 4.39 (d, 1H, $J = 6.2$ Hz), 3.77 (s, 3H), 3.63–3.55 (m, 1H), 2.73 (d, 2H, $J = 7.2$ Hz), 2.39 (s, 3H), 1.21 (d, 3H, $J = 6.5$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 158.9, 142.8, 137.4, 131.6, 129.4, 128.9, 126.9, 124.7, 118.1, 113.5, 55.4, 50.5, 42.3, 22.3, 21.4; IR ν_{max} 2918, 1378, 1199 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{17}\text{H}_{19}\text{BrNO}_3\text{S}$ [$\text{M} - \text{H}$] $^-$ HRMS(ESI): 396.0275; found: 396.0261.

1-(2-Trimethoxyphenyl)-2-tosylamino-propane and 1-(4-trimethoxyphenyl)-2-tosylamino-propane (12a + 12b). Anisole (0.53 mL); tosylamide (342.4 mg); FCC–DCM/MeOH (9.5:0.5). **12a + 12b** (236.2 mg, 74%, isomeric ratio after purification: 1.3/1); light yellow oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) compound **12a** δ 7.45 (d, 2H, $J = 8.2$ Hz), 7.16 (t, 1H, $J = 7.6$ Hz), 7.09 (d, 2H, $J = 8.0$ Hz), 6.89 (d, 1H, $J = 7.4$ Hz), 6.79 (t, 1H, $J = 7.3$ Hz), 6.72 (d, 1H, $J = 7.6$ Hz), 4.83 (d, 1H, $J = 5.7$ Hz), 3.67 (s, 3H), 3.45–3.34 (m, 1H), 2.71 (dd, 1H, $J = 13.6, 8.6$ Hz), 2.63–2.65 (m, 1H), 2.29 (s, 3H), 1.14 (d, 3H, $J = 6.4$ Hz); compound **12b** δ 7.61 (d, 2H, $J = 8.2$ Hz), 7.22 (d, 2H, $J = 7.9$ Hz), 6.92 (d, 2H, $J = 8.4$ Hz), 6.75 (d, 1H, $J = 8.4$ Hz), 4.20 (d, 1H, $J = 7.0$ Hz), 3.71 (s, 3H), 3.45–3.34 (m, 1H), 2.63–2.56 (m, 2H), 2.34 (s, 3H), 1.01 (d, 3H, $J = 6.5$ Hz). The characterization of product **12a** is consistent with that reported in the literature.¹⁹ The characterization of product **12b** is consistent with that reported in the literature.²⁰

1-(3-Thienyl)-2-tosylamino-propane (13b). Thiophene (0.40 mL); tosylamide (342.4 mg); FCC–DCM/MeOH (9.5:0.5). **13b** (115.1 mg, 39%); light yellow oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 7.66 (d, 2H, $J = 7.9$ Hz), 7.26 (d, 2H, $J = 8.0$ Hz), 7.19 (dd, 1H, $J = 4.9, 3.0$ Hz), 6.88 (d, 1H, $J = 2.1$ Hz), 6.76 (d, 1H, $J = 4.6$ Hz), 4.25 (d, 1H, $J = 6.4$ Hz), 3.57–3.50 (m, 1H), 2.71 (d, 2H, $J = 5.1$ Hz), 2.42 (s, 3H), 1.09 (d, 3H, $J = 6.5$ Hz). The characterization of product **13b** is consistent with that reported in the literature.²¹

General Procedure for the Synthesis of Arylated/Hydroaminated Products with Different Substituted Alcohols. In a sealed tube, after 15 min, the crotyl alcohol or 1-buten-3-ol (1.0 mmol, 72 mg), hydrocarbons (5.0 mmol), and sulfonamide (2.0 mmol) were added to a solution of $\text{Cu}(\text{OTf})_2$ (10 mol %, 36.2 mg) and xantphos (20 mol %, 11.6 mg) in chlorobenzene (0.25 M). The resulted solution was magnetically stirred and heated at 75 °C in an oil bath for 24 h. The reaction mixture was washed with brine (3 \times 5 mL) and the organic layer was extracted with AcOEt (2 \times 5 mL), dried over MgSO_4 , and filtered. The solvent was evaporated under reduced pressure. The residue was purified by FCC. Starting from the aromatic source and appropriate sulfonamide, yield, physical, spectroscopic, and analytical data of compounds **14a–b**, **15a–b** and **16–17** are as follows.

1-(2,4,6-Trimethylphenyl)-2-(benzenesulfonamido)-butane and 1-(2,4,6-trimethylphenyl)-3-(benzenesulfonamido)-butane (14a + 14b). Mesitylene (0.69 mL); benzenesulfonamide (314.4 mg); FCC–MeOH/DCM (1:50), R_f : 0.37. **14a + 14b** (218.6 mg, 66%, isomeric ratio after purification: 1/1.4); colorless oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) compound **14a** δ 7.63 (d, 2H, $J = 8.8$ Hz), 7.52–7.48 (m, 2H), 7.36 (t, 1H, $J = 7.8$ Hz), 6.70 (s, 2H), 4.33 (d, 1H, $J = 7.7$ Hz), 3.39–3.33 (m, 1H), 2.78 (dd, 1H, $J = 14.1, 7.8$ Hz), 2.68 (dd, 1H, $J = 14.2, 7.6$ Hz), 2.22 (s, 3H), 2.15 (s, 6H), 1.66–1.38 (m, 2H), 0.84 (t, 3H, $J = 7.3$ Hz); compound **14b** δ 7.92 (d, 2H, $J = 9.0$ Hz), 7.52–7.48 (m, 2H), 7.36 (t, 1H, $J = 7.8$ Hz), 6.80 (s, 2H), 4.38 (d, 1H, $J = 8.7$ Hz),

3.50–3.43 (m, 1H), 2.58–2.49 (m, 2H), 2.23 (s, 3H), 2.18 (s, 6H), 1.66–1.39 (m, 2H), 1.14 (d, 3H, $J = 6.6$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 141.2, 140.5, 136.5, 135.7, 135.6, 135.2, 134.9, 132.6, 132.1, 131.4, 129.3, 129.1, 128.9, 128.7, 126.9, 126.8, 55.6, 50.7, 36.8, 34.8, 28.2, 25.5, 21.8, 20.77, 20.75, 20.3, 19.6, 10.0; IR ν_{max} 2965, 1322, 1158 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{19}\text{H}_{24}\text{NO}_2\text{S}$ [$\text{M} - \text{H}$] $^-$: 330.1533; found: 330.1531.

1-(2,4,6-Trimethylphenyl)-2-(4-chlorobenzenesulfonamido)-butane and 1-(2,4,6-trimethylphenyl)-3-(4-chlorobenzenesulfonamido)-butane (15a + 15b). Mesitylene (0.69 mL); 4-chlorobenzenesulfonamide (383.3 mg); FCC–DCM, R_f : 0.34. **15a + 15b** (222.7 mg, 61%, isomeric ratio after purification: 1/1); colorless oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) compound **15a** δ 7.73 (d, 2H, $J = 8.5$ Hz), 7.44–7.44 (m, 2H), 6.90 (s, 2H), 4.31 (d, 1H, $J = 7.8$ Hz), 3.41–3.34 (m, 1H), 2.75–2.66 (m, 2H), 2.31 (s, 3H), 2.14 (s, 6H), 1.67–1.57 (m, 2H), 0.93 (t, 3H, $J = 7.9$ Hz); compound **15b** δ 7.84 (d, 2H, $J = 8.4$ Hz), 7.49 (d, 2H, $J = 8.4$ Hz), 6.81 (s, 2H), 4.37 (d, 1H, $J = 8.3$ Hz), 3.49–3.42 (m, 1H), 2.53–2.51 (m, 1H), 2.49–2.40 (m, 1H), 2.24 (s, 3H), 2.19 (s, 6H), 1.54–1.40 (m, 2H), 1.16 (d, 3H, $J = 6.6$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 143.7, 142.1, 140.1, 139.0, 136.3, 136.0, 135.6, 135.3, 134.8, 132.3, 131.3, 129.4, 128.9, 128.8, 128.0, 127.7, 55.9, 50.8, 36.8, 34.4, 29.1, 25.5, 22.8, 21.7, 20.8, 20.3, 19.5, 10.1; IR ν_{max} 2919, 1314, 1155 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{19}\text{H}_{23}\text{ClNO}_2\text{S}$ [$\text{M} - \text{H}$] $^-$: 364.1144; found: 364.1128.

1-(2,4,6-Trimethylphenyl)-2-(o-nosylamino)-butane (16). Mesitylene (0.69 mL); 4-nitrobenzenesulfonamide (404.4 mg); FCC–AcOEt/PE (1:4), R_f : 0.33. **16** (236.9 mg, 63%) yellow oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 7.82–7.76 (m, 2H), 7.64–7.54 (m, 2H), 6.54 (s, 2H), 5.26 (d, 1H, $J = 7.9$ Hz), 3.80–3.71 (m, 1H), 2.80 (dd, 1H, $J = 14.4, 8.8$ Hz), 2.75 (dd, 1H, $J = 14.4, 6.9$ Hz), 2.18 (s, 6H), 2.12 (s, 3H), 1.72–1.58 (m, 2H), 0.97 (t, 3H, $J = 7.4$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 101 MHz) δ 144.4, 136.5, 135.5, 132.8, 132.4, 131.2, 129.9, 129.1 (2C), 125.3, 56.9, 34.8, 29.3, 20.7, 20.4, 10.3; IR ν_{max} 2916, 1356, 1164 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{19}\text{H}_{23}\text{N}_2\text{O}_4\text{S}$ [$\text{M} - \text{H}$] $^-$: 375.1384; found: 375.1382.

1-(2,4,6-Trimethylphenyl)-2-(p-nosylamino)-butane (17). Mesitylene (0.69 mL); 4-nitrobenzenesulfonamide (404.4 mg); FCC–AcOEt/PE (1:4), R_f : 0.33. **17** (255.8 mg, 68%) yellow oil. $^1\text{H NMR}$ (CDCl_3 , 400 MHz) 8.08 (d, 2H, $J = 8.4$ Hz), 7.62 (d, 2H, $J = 8.2$ Hz), 6.59 (s, 2H), 4.35 (d, 1H, $J = 7.8$ Hz), 3.51–3.49 (m, 1H), 2.68 (d, 2H, $J = 7.6$ Hz), 2.18 (s, 3H), 2.12 (s, 6H), 1.73–1.66 (m, 2H), 1.00 (t, 3H, $J = 7.5$ Hz); ^{13}C NMR (CDCl_3 , 101 MHz) δ 150.5, 146.2, 136.3, 136.2, 131.2, 129.2, 127.5, 123.7, 56.6, 34.2, 29.9, 20.6, 20.3, 10.1; IR ν_{max} 2918, 1347, 1155 cm^{-1} . HRMS(ESI): m/z calc. for $\text{C}_{19}\text{H}_{23}\text{N}_2\text{O}_4\text{S}$ [$\text{M} - \text{H}$] $^-$: 375.1384; found: 375.1379.

ASSOCIATED CONTENT

Data Availability Statement

The data underlying this study are available in the published article and its online [Supporting Materials](#).

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.joc.3c01536>.

^1H and ^{13}C NMR spectra of compounds **1a–h**, **3a–f**, **4a–d**, **5a–c**, **6–11**, **12a–b**, **13a–b**, **14a–b**, **15a–b**, **16** and **17** (PDF)

AUTHOR INFORMATION

Corresponding Authors

Camilla Loro – Dipartimento di Scienza e Alta Tecnologia, Università dell'Insubria, 22100 Como, Italy; orcid.org/0000-0001-9616-2335; Email: camilla.loro@uninsubria.it
Julie Oble – Faculté des Sciences et Ingénierie, CNRS, Institut Parisien de Chimie Moléculaire, IPCM, Sorbonne Université, 75005 Paris, France; orcid.org/0000-0002-4002-255X; Email: julie.oble@sorbonne-universite.fr

Authors

Marta Papis – Dipartimento di Scienza e Alta Tecnologia, Università dell'Insubria, 22100 Como, Italy

Francesca Foschi – Dipartimento di Scienza e Alta Tecnologia, Università dell'Insubria, 22100 Como, Italy

Gianluigi Broggini – Dipartimento di Scienza e Alta Tecnologia, Università dell'Insubria, 22100 Como, Italy; orcid.org/0000-0003-2492-5078

Giovanni Poli – Faculté des Sciences et Ingénierie, CNRS, Institut Parisien de Chimie Moléculaire, IPCM, Sorbonne Université, 75005 Paris, France; orcid.org/0000-0002-7356-1568

Complete contact information is available at:

<https://pubs.acs.org/10.1021/acs.joc.3c01536>

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

C.L., M.P., F.F., and G.B. thank Università degli Studi dell'Insubria for financial support. J.O. and G.P. acknowledge support by Sorbonne Université and CNRS.

REFERENCES

- (1) (a) Pelletier, S. W. *Chemistry of Alkaloids*; Van Nostrand Reinhold: New York, 1970. (b) Leake, C. D. *The Amphetamines: Their Actions and Uses*; Charles C. Thomas Co.: Springfield, 1958. (c) Testa, B.; Salvesen, B. J. Quantitative structure-activity relationships in drug metabolism and disposition. Pharmacokinetics of *N*-substituted amphetamines in humans. *J. Pharm. Sci.* **1980**, *69*, 497–501. (d) Jian-min, C.; Zhi-yuan, W.; Shi-xuan, W.; Rui, S.; Rui, S.; Ning, W.; Ning, W.; Jin, L. Effects of Lisdexamfetamine, a Prodrug of *D*-Amphetamine, on Locomotion, Spatial Cognitive Processing and Neurochemical Profiles in Rats: A Comparison With Immediate-Release Amphetamine. *Front. Psychiatry* **2022**, *13*, No. 885574. (e) Nieddu, M.; Baralla, E.; Pasciu, V.; Rimoli, M. G.; Boatto, G. Cross-reactivity of commercial immunoassays for screening of new amphetamine designer drugs. A review. *J. Pharm. Biomed. Anal.* **2022**, *218*, No. 114868.
- (2) (a) Imm, S.; Bahn, S.; Neubert, L.; Neumann, H.; Beller, M. An Efficient and General Synthesis of Primary Amines by Ruthenium-Catalyzed Amination of Secondary Alcohols with Ammonia. *Angew. Chem., Int. Ed.* **2010**, *49*, 8126–8129. (b) Muñoz, L.; Rodríguez, A. M.; Rosell, G.; Bosch, M. P.; Guerrero, A. Enzymatic enantiomeric resolution of phenylethylamines structurally related to amphetamine. *Org. Biomol. Chem.* **2011**, *9*, 8171–8177. (c) Jagadeesh, R. V.; Murugesan, K.; Alshammari, A. S.; Neumann, H.; Pohl, M.-M.; Radnik, J.; Beller, M. MOF-Derived cobalt nanoparticles catalyze a general synthesis of amine. *Science* **2017**, *358*, 326–332. (d) González-Martínez, D.; Gotor, V.; Gotor-Fernández, V. Stereoselective Synthesis of 1-Arylpropan-2-amines from Allylbenzenes through a Wacker-Tsuji Oxidation-Biotransamination Sequential Process. *Adv. Synth. Catal.* **2019**, *361*, 2582–2593. (e) Albarrán-Velo, J.; Gotor-Fernández, V.; Lavandera, I. Markovnikov Wacker-Tsuji Oxidation of Allyl(hetero)arenes and Application in a One-Pot Photo-Metal-Biocatalytic Approach to Enantioenriched Amines and Alcohols. *Adv. Synth. Catal.* **2021**, *363*, 4096–4108.
- (3) (a) Griffith, R. C.; Gentile, R. J.; Davidson, T. A.; Scott, F. L. Convenient One-Step Synthesis of *N*-Substituted α -Methylphenylamines via Aminomercuration-Demercuration. *J. Org. Chem.* **1979**, *44*, 3580–3583. (b) Hartung, C. G.; Breindl, C.; Tillack, A.; Beller, M. A Base-Catalyzed Domino-Isomerization-Hydroamination Reaction - A New Synthetic Route to Amphetamines. *Tetrahedron* **2000**, *56*, 5157–5162. (c) Utsunomiya, M.; Hartwig, J. F. Ruthenium-catalyzed anti-markovnikov hydroamination of vinylarenes. *J. Am. Chem. Soc.* **2004**, *126*, 2702–2703. (d) Jaspers, D.; Kubiak, R.; Doye, S. Recyclable Gallium as Catalyst Precursor for a Convenient and Solvent-Free Method for the Intermolecular Addition of Sulfonamides to Alkenes. *Synlett* **2010**, *8*, 1268–1272. (e) Giner, X.; Nájera, C.; Kovács, G.; Lledós, A.; Ujaque, G. Gold versus Silver-Catalyzed Intermolecular Hydroaminations of Alkenes and Dienes. *Adv. Synth. Catal.* **2011**, *353*, 3451–3466.
- (4) (a) Foschi, F.; Loro, C.; Sala, R.; Oble, J.; Lo Presti, L.; Beccalli, E. M.; Poli, G.; Broggini, G. Intramolecular Aminoacidation of Unactivated Terminal Alkenes by Palladium-Catalyzed Reactions with Hydrogen Peroxide as the Oxidant. *Org. Lett.* **2020**, *22*, 1402–1406. (b) Giofrè, S.; Loro, C.; Molteni, L.; Castellano, C.; Contini, A.; Nava, D.; Broggini, G.; Beccalli, E. M. Copper(II)-Catalyzed Aminohalogenation of Alkynyl Carbamates. *Eur. J. Org. Chem.* **2021**, *2021*, 1750–1757. (c) Loro, C.; Molteni, L.; Papis, M.; Beccalli, E. M.; Nava, D.; Lo Presti, L.; Brenna, S.; Colombo, G.; Foschi, F.; Broggini, G. Direct Synthesis of Fluorescent Oxazolophenoxazines by Copper-Catalyzed/Hypervalent Iodine(III)-Mediated Dimerization/Cyclization of 2-Benzylamino-phenols. *J. Org. Chem.* **2022**, *87*, 1032–1042. (5) Loro, C.; Oble, J.; Foschi, F.; Papis, M.; Beccalli, E. M.; Giofrè, S.; Poli, G.; Broggini, G. Acid-mediated decarboxylative C-H coupling between arenes and *O*-allyl carbamates. *Org. Chem. Front.* **2022**, *9*, 1711–1718. (6) (a) Kimura, M.; Futamata, M.; Mukai, R.; Tamaro, Y. Pd-catalyzed C3-selective allylation of indoles with allyl alcohols promoted by triethylborane. *J. Am. Chem. Soc.* **2005**, *127*, 4592–4593. (b) Usui, I.; Schmidt, S.; Keller, M.; Breit, B. Allylation of *N*-Heterocycles with Allylic Alcohols Employing Self-Assembling Palladium Phosphane Catalysts. *Org. Lett.* **2008**, *10*, 1207–1210. (c) Zaitsev, A. B.; Gruber, S.; Plüss, P. A.; Pregosin, P. S.; Veiros, L. F.; Würle, M. Fast and highly regioselective allylation of indole and pyrrole compounds by allyl alcohols using Ru-sulfonate catalysts. *J. Am. Chem. Soc.* **2008**, *130*, 11604–11605. (d) Mukherjee, P.; Widenhofer, R. A. Gold(I)-Catalyzed Enantioselective Intramolecular Dehydrative Amination of Allylic Alcohols with Carbamates. *Angew. Chem., Int. Ed.* **2012**, *51*, 1405–1407. (e) Huang, W.-Y.; Nishikawa, T.; Nakazaki, A. Palladium-Catalyzed Cascade Wacker/Allylation Sequence with Allyl Alcohols Leading to Allylated Dihydropyrones. *ACS Omega* **2017**, *2*, 487–495. (7) (a) Sanz, R.; Martínez, A.; Miguel, D.; Álvarez-Gutiérrez, J. M.; Rodríguez, F. Brønsted Acid-Catalyzed Nucleophilic Substitution of Alcohols. *Adv. Synth. Catal.* **2006**, *348*, 1841–1845. (b) Trillo, P.; Baeza, A.; Nájera, C. Fluorinated Alcohols As Promoters for the Metal-Free Direct Substitution Reaction of Allylic Alcohols with Nitrogenated, Silylated, and Carbon Nucleophiles. *J. Org. Chem.* **2012**, *77*, 7344–7354. (c) Trillo, P.; Baeza, A.; Nájera, C. FeCl₃·6H₂O and TfOH as Catalysts for Allylic Amination Reaction: A Comparative Study. *Eur. J. Org. Chem.* **2012**, *2012*, 2929–2934. (d) Zhuang, M.; Du, H. Chiral Brønsted acid catalyzed enantioselective intermolecular allylic aminations. *Org. Biomol. Chem.* **2014**, *12*, 4590–4593. (e) Luo, Z.; Liu, Z.-Q.; Yang, T.-T.; Zhuang, X.; Hong, C.-M.; Zou, F.-F.; Xue, Z.-Y.; Li, Q.-H.; Liu, T.-L. 1,1,1,3,3,3-Hexafluoro-2-propanol (HFIP)-Assisted Catalyst-Free Sulfonation of Allylic Alcohols with Sulfinyl Amides. *Org. Lett.* **2022**, *24*, 741–745. (8) (a) Rao, W.; Chan, P. W. H. Gold-catalyzed allylic alkylation of aromatic and heteroaromatic compounds with allylic alcohols. *Org. Biomol. Chem.* **2008**, *6*, 2426–2433. (b) Namba, K.; Yamamoto, H.; Sasaki, I.; Mori, K.; Imagawa, H.; Nishizawa, M. Hg(OTf)₂-Catalyzed Arylene Cyclization. *Org. Lett.* **2008**, *10*, 1767–1770. (c) Yamamoto, Y.; Itonaga, K. Synthesis of Chromans via [3 + 3] Cyclocoupling of Phenols with Allylic Alcohols Using a Mo/*o*-Chloranil Catalyst System. *Org. Lett.* **2009**, *11*, 717–720. (d) Kazakova, A. N.; Iakovenko, R. O.; Boyarskaya, I. A.; Ivanov, A. Y.; Avdontceva, M. S.; Zolotarev, A. A.; Panikarovskiy, T. L.; Starova, G. L.; Nenajdenko, V. G.; Vasilyev, A. V. Brominated CF₃-allyl alcohols as multivalent electrophiles in TfOH promoted reactions with arenes. *Org. Chem. Front.* **2017**, *4*, 255–265. (e) Vu, M. D.; Qing Foo, C.; Sadeer, A.; Shand, S. S.; Li, Y.; Pullarkat, S. A. Triflic-Acid-Catalyzed Tandem Allylic Substitution-Cyclization Reaction of Alcohols with Thiophe-

nols-Facile Access to Polysubstituted Thiochromans. *ACS Omega* **2018**, *3*, 8945–8951.

(9) McCubbin, J. A.; Hosseini, H.; Krokhn, O. V. Boronic Acid Catalyzed Friedel-Crafts Reactions of Allylic Alcohols with Electron-Rich Arenes and Heteroarenes. *J. Org. Chem.* **2010**, *75*, 959–962.

(10) Zhou, X.; Zhang, H.; Xie, X.; Li, Z. Efficient Synthesis of 3-Iodoindenes via Lewis-Acid Catalyzed Friedel-Crafts Cyclization of Iodinated Allyl Alcohols. *J. Org. Chem.* **2008**, *73*, 3958–3960.

(11) Bandini, M.; Eichholzer, A.; Kotrusz, P.; Tragni, M.; Troisi, S.; Umami-Ronchi, A. Ligand-Free Silver(I)-Catalyzed Intramolecular Friedel-Crafts Alkylation of Arenes with Allyl Alcohols. *Adv. Synth. Catal.* **2009**, *351*, 319–324.

(12) Other mono and bidentate ligands, such as PPh₃, tri(2-furyl)-phosphine, BINAP, dppe, phanephos, (S,S)-(-)-2,2'-isopropylidene-bis(4-*t*-butyl-2-oxazoline) and (S)-2-benzyl-4-phenyl-2-oxazoline were tested. However, only phosphine ligands could lead to compound **1a**, although in lower yields than xantphos.

(13) A gram-scale synthesis was carried on 5.0 mmol of allyl alcohol, 25.0 mmol of mesitylene and 10.0 mmol of tosyl amide at 75 °C for 48 h. The 1-(2,4,6-trimethylphenyl)2-tosylamino-propane **2a** was obtained with 61% yield (1.01 g).

(14) (a) Tschan, M.; Thomas, C. M.; Strub, H.; Carpentier, J.-F. Copper(II) triflate as source of triflic acid: effective, green catalyst of hydroalkoxylation reactions. *Adv. Synth. Catal.* **2009**, *351*, 2496–2505. (b) Michon, C.; Medina, F.; Capet, F.; Roussel, R.; Agbossou-Niedercorn, F. Inter- and intramolecular hydroamination of unactivated alkenes catalyzed by combination of copper and silver salts: the unveiling of Brønsted acid catalyst. *Adv. Synth. Catal.* **2010**, *352*, 3293–3305. (c) Taylor, J. G.; Adrio, L. A.; Hii, K. K. Hydroamination reactions by metal triflates: Brønsted acid vs. metal catalysis? *Dalton Trans.* **2010**, *39*, 1171–1175. (d) Dang, T. T.; Boeck, F.; Hinterman, L. Hidden Brønsted acid catalyst pathways of accidental or deliberate generation of triflic acid from metal triflates. *J. Org. Chem.* **2011**, *76*, 9353–9361.

(15) (a) Taylor, J. G.; Whittall, N.; Hii, K. K. Copper-catalyzed intermolecular hydroamination of alkenes. *Org. Lett.* **2006**, *8*, 3561–3564. (b) Rao, W.; Kothandaraman, P.; Boon Koh, C.; Hong Chan, P. W. Copper(II) Triflate-Catalyzed Intramolecular Hydroamination of Homoallylic Amino Alcohols as an Expedient Route to *trans*-2,5-Dihydro-1*H*-pyrroles and 1,2-Dihydroquinolines. *Adv. Synth. Catal.* **2010**, *352*, 2521–2530.

(16) The use of more substituted alcohols (such as 2-penten-1-ol) lead to extremely complex mixtures.

(17) (a) Prakash, G. K. S.; Mathew, T.; Panja, C.; Kulkarni, A.; Olah, G. A.; Harmer, M. A. Tetrafluoroethanesulfonic acid, HC₂F₄SO₃H and gallium tetrafluoroethanesulfonate as effective catalysts in organic synthesis. *Adv. Synth. Catal.* **2012**, *354*, 2163–2171. (b) Zhang, Y.; Zhang, Y.; Chen, L.; Lu, T. A copper(II) triflate-catalyzed tandem Friedel-Crafts alkylation/cyclization process towards dihydroindenes. *Adv. Synth. Catal.* **2011**, *353*, 1055–1060.

(18) The reason why benzenesulfonamide and 4-chlorobenzenesulfonamide afford a nearly random mixture of the regioisomers, while 2-nitrobenzenesulfonamide and 4-nitrobenzenesulfonamide give only one regioisomer is at present not clear. A possible rationalization is as follows: on the one hand benzenesulfonamide and 4-chlorobenzenesulfonamide may rapidly intercept the kinetically and randomly generated carbenium ions **VII** and **VIII**, to give a mixture of the two observed regioisomers; on the other hand, cation interception by the less nucleophilic nitrobenzene sulfonamides may take place only after a 1,2-hydride shift mediated carbenium ion equilibration toward the seemingly more stable carbenium ion **VIII**. However, as in our previous study, a homobenzylic-to-benzylic 1,2-hydride shift is not observed.

(19) Nenajdenko, V. G.; Karpov, A. S.; Balenkova, E. S. A new convenient approach to chiral β -aryl(heteroaryl)alkylamines. *Tetrahedron Asym.* **2001**, *12*, 2517–2527.

(20) Ziegler, D. S.; Karaghiosoff, K.; Knochel, P. Generation of Aryl and Heteroaryl Magnesium Reagents in Toluene by Br/Mg or Cl/Mg Exchange. *Angew. Chem., Int. Ed.* **2018**, *57*, 6701–6704.

(21) Perrey, D.; Zhang, D.; Nguyen, T.; Carroll, F. I.; Ko, M.-C.; Zhang, Y. Synthesis of Enantiopure PZM21: A Biased Agonist of the Mu-Opioid Receptor. *Eur. J. Org. Chem.* **2018**, *2018*, 4006–4012.

Recommended by ACS

Cobalt-Promoted Electroreductive Cross-Coupling of Prop-2-yn-1-yl Acetates with Chloro(vinyl)silanes

Chong-Hui Xu, Jin-Heng Li, *et al.*

SEPTEMBER 27, 2023
ORGANIC LETTERS

READ 

Rh-Catalyzed [2,3]-Sigmatropic Rearrangement of Alkynyl Carbenes with Allyl Sulfides to Access Sulfide-Substituted 1,5-Enynes

Zhongxue Fang, Yongquan Ning, *et al.*

JUNE 30, 2023
THE JOURNAL OF ORGANIC CHEMISTRY

READ 

Dual Nickel/Photoredox-Catalyzed Asymmetric Carbonylation of Alkenes

Xiaoyong Du, Cristina Nevado, *et al.*

MAY 30, 2023
JOURNAL OF THE AMERICAN CHEMICAL SOCIETY

READ 

Cobalt(III)-Catalyzed and DMSO-Involved Allylation of 1,3-Dicarbonyl Compounds with Alkenes

Xuefeng Xu, Xu Zhang, *et al.*

OCTOBER 20, 2022
THE JOURNAL OF ORGANIC CHEMISTRY

READ 

Get More Suggestions >